

A Sequential Metal-Catalyzed C–N Bond Formation in the Synthesis of 2-Amido-indoles

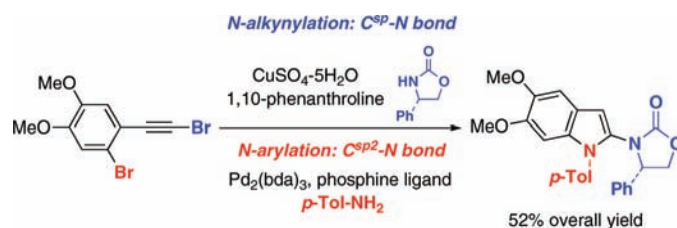
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ABSTRACT



A sequential metal-catalyzed C–N bond formation employing *ortho*-haloaryl acetylenic bromides is described. The initial amidation is highly selective for C^{sp}–N bond formation, leading to *o*-haloaryl-substituted ynamides that can be useful building blocks, while the overall sequence provides a facile construction of 2-amido-indoles.

Given the importance of heterocyclic manifolds,¹ we have been developing synthetic methods that feature ynamides^{2,3} en route to various heterocycles.⁴ These efforts led us to examine a possible entry for constructing amide-substituted

indoles.^{5–7} Specifically, as shown in Scheme 1, this pathway would commence with *ortho*-haloaryl acetylenic bromides **1** and adopt a consecutive metal-catalyzed C–N bond

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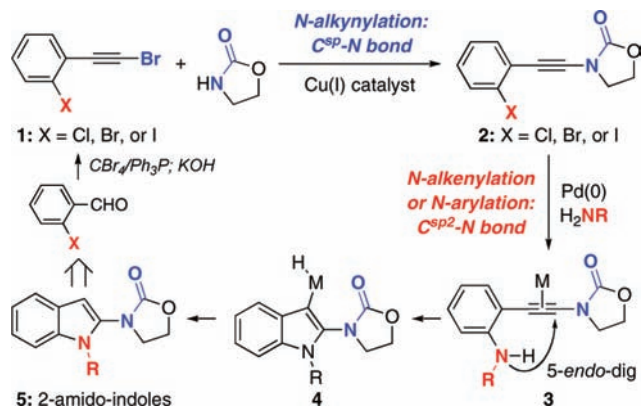
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Scheme 1



formation⁸ with the first involving the sp -hybridized carbon^{9–13} in an N-alkynylation manner and the second one pertaining to an sp^2 -hybridized carbon in a N-arylation manner.^{8,14} The second C–N bond formation can also occur in a tandem manner with the ensuing indole formation promoted by the metal^{15,16} in a 5-endo-dig cyclization mode via **3**. While copper can be employed to catalyze the C^{sp} –N formation,^{9–13} we intend to utilize palladium for the C^{sp2} –N formation.¹⁵ If this sequential C–N bond formation is selective, it would constitute a facile entry to de novo

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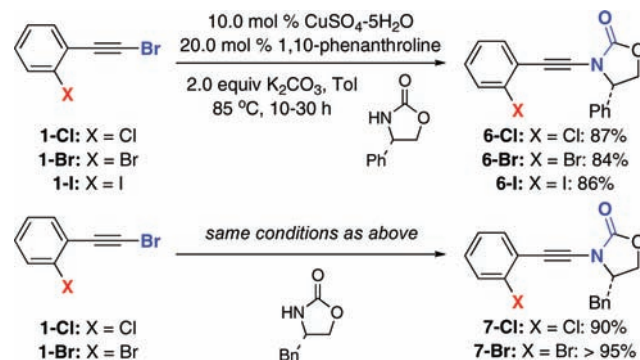
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2-amido-indoles^{17,18} **5** from *ortho*-haloaryl acetylenes **1**, which can be readily derived from aromatic aldehydes in two steps.¹⁹ We report here the synthesis of 2-amido-indoles via a sequential metal-catalyzed C–N bond formation.

A selective amidative cross-coupling of *ortho*-haloaryl acetylenic bromides **1**²⁰ could be readily established as shown in Scheme 2. By employing 10 mol % of $CuSO_4 \cdot 5H_2O$ and

Scheme 2



20 mol % of 1,10-phenanthroline,¹³ ynamide **6-Cl** was attained in 87% yield from **1-Cl**. The amidation remained selective when using **1-Br** and even **1-I**, leading to **6-Br** and **6-I** in 84% and 86% yield, respectively. Under the same conditions, ynamides **7-Cl** and **7-Br** were obtained also via a highly selective C^{sp} –N formation.

A diverse array of *ortho*-haloaryl acetylenic bromides could be subjected to this selective amidation to give ynamides **8–16** (Figure 1). Moreover, a range of cyclic and acyclic amides including sulfonamides could be employed for the N-alkynylation to afford ynamides **17–23** in good yields.

Having established this selective amidation, we recognized that we have an excellent protocol to access *o*-haloaryl-

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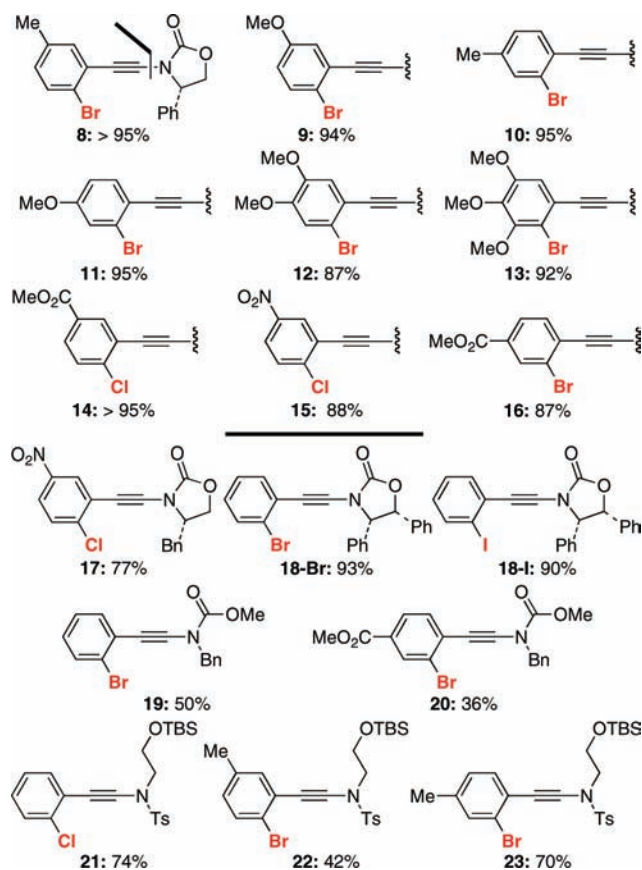


Figure 1. *N*-Alkynylation products. Reaction conditions are the same as those in Scheme 2. All are isolated yields.

substituted ynamides **6–23**, which represent a new class of functionally rich building blocks that could be utilized in a number of transformations involving either the *o*-haloaryl or ynamido motif (Figure 2), leading to rapid assembly of structural complexity.

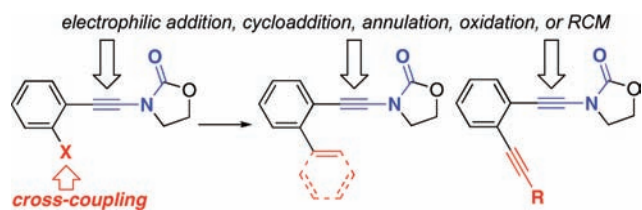
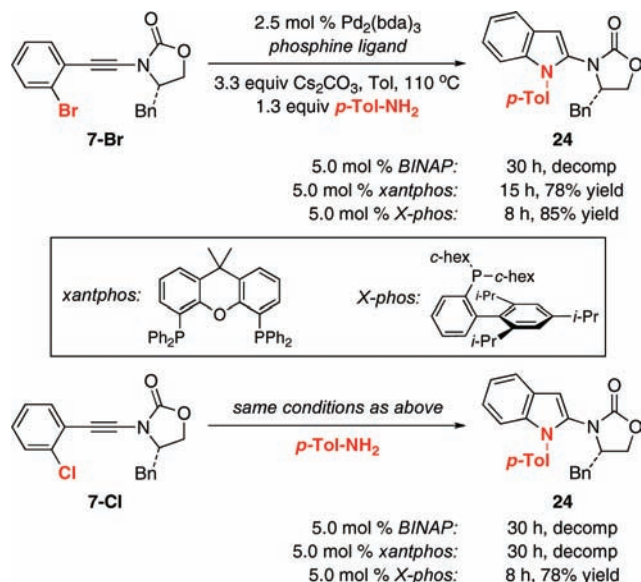


Figure 2. Synthetic potential of *o*-haloaryl ynamides.

To illustrate such synthetic potential, we chose to pursue aminative cross-coupling of aryl halides^{8,14} to access 2-acetylenic anilines en route to 2-amido-indoles via metal-promoted 5-*endo*-dig cyclization.^{15–19,21} As shown in Scheme 3, when ynamide **7-Br** was subjected to amination conditions employing 2.5 mol % of Pd₂(dba)₃ and *p*-Tol-NH₂, 2-amido-indole **24** was obtained in good yields when using either

Scheme 3



5.0 mol % of van Leeuwen's xantphos²² or Buchwald's X-phos as ligands.²³ Intriguingly, the use of X-phos appears to shorten the reaction time relative to xantphos, while BINAP was not useful. On the other hand, amination of **7-Cl** led to **24** in 78% yield only when using X-phos. It is noteworthy that amination of **6-I** gave only 26% yield of the corresponding indole (not shown), thereby suggesting that aryl chlorides and bromides are better suited in this operation than aryl iodides.

The generality of this tandem amination-5-*endo*-dig cyclization is shown in Table 1, featuring a range of different amines and *o*-chloroaryl- or *o*-bromoaryl-substituted ynamides in excellent yields for their respective reactions. X-ray crystallographic analysis of 2-amido-indole **28** reveals unique orthogonality of three planes: 2-oxazolidone, the indole ring, and the *para*-tolyl ring (Figure 3). Structures with related orthogonality have been shown²⁴ to possess inhibitory

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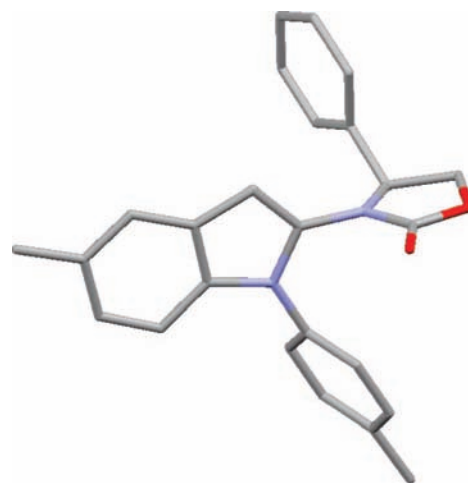
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Table 1. N-Alkenylation in the 2-Amido-indole Synthesis

entry	ynamides ^{a,b}	2-amido-indoles	yield [%] ^c
1		6-Br	25: 91
2		6-Br	26: 82
3		7-Cl	27^d: 91
4		8	28: 65
5		12	29: 60
6		14	30: 72
7		15	31: 64
8		16	32: 80
9		18-Br	33: 71
10		20	34: 82
11		20	35: 75
12		21	36: 88

^a Reaction conditions: 2.5 mol % of Pd₂(bda)₃, 5.0 mol % of X-phos, 3.3 equiv of Cs₂CO₃, 1.3 equiv of R-NH₂, 110 °C, 8–24 h. ^b Toluene was used as solvent in entries 4, 5, 7, and 9, and dioxane was used in entries 1–3, 6, 8, and 10–12. ^c Isolated yields. ^d PMP = *para*-methoxy-phenyl.

activities against human peptidyl prolyl *cis/trans* isomerase [PPI] Pin-1,²⁵ which catalyzes the isomerization of prolyl

**Figure 3.** X-Ray structure of 2-amido-indole **28**.

peptides from *cis* to *trans*^{26,27} and accommodates such orthogonality at its active site. We are currently investigating such potential biological activity.

We have described here a sequential metal-catalyzed C–N bond formation employing *ortho*-haloaryl acetylenic bromides. The initial amidation is highly selective for the C^{sp}–N bond, leading to *o*-haloaryl-substituted ynamides that can be useful building blocks. The overall sequence provides a facile construction of 2-amido-indoles possessing a unique structural manifold.

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Supporting Information Available: Experimental procedures as well as NMR spectra, characterizations, and X-ray crystallographic data. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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